ACETYLVISMIONE D FROM PSOROSPERMUM FEBRIFUGUM*

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(Received 10 July 1985)

Key Word Index-Psorospermum febrifugum; Guttiferae; acetylvismione D.

Abstract—An acetone extract of the root-bark of *Psorospermum febrifugum*, collected in Zaire, afforded the known geranyloxyemodin, bianthrone A1, vismiones D and F, and the new acetylvismione D.

INTRODUCTION

The genus *Psorospermum* (tribe Vismieae) is a source of vismiones and other biogenetically linked anthranoids, which show cytotoxic activity and feeding deterrence [1]. Anthranoids from *Psorospermum* spp. are characterized by the presence of a geranyl substituent. This paper reports on the constituents of the root-bark of *Psorospermum febrifugum*, collected in Zaire.

RESULTS AND DISCUSSION

The cold acetone extract of the root-bark of P. febrifugum gave geranyloxyemodin [2], bianthrone A1 [3] and vismiones D (1) [2] and F [3]. A new compound (2), $C_{27}H_{32}O_6$, showed the typical spectral features of an acetylvismione [2]. Its ¹H NMR spectrum disclosed also the presence of three aromatic protons, two of which were meta coupled, and of an O-geranyl side chain. The compound was thus assigned structure 2 and given the name acetylvismione D. In the mass spectrum, the losses of acetic acid and $C_{10}H_{16}$ gave a base peak at m/z 256, analogous to that given by vismione D; the rest of the spectrum was superimposable on that of emodin anthrone [2]. On silica gel, acetylvismione D gave geranyloxyemodin, bianthrone A1 and emodin, as did vismione D [2].

EXPERIMENTAL

Roots of *Psorospermum febrifugum* Spach were collected in Zaire and identified at the University of Kinshasa. A voucher specimen is deposited in the Herbarium of Centro Chimica dei Recettori under the cypher PFKK. The root-bark (76 g) was extracted with cold Me₂CO (3×300 ml) for 45 days to give a residue of 9.54 g. CC of a part of the extract (7.8 g) on silica gel with CHCl₃-MeOH mixtures afforded: geranyloxyemodin [2] (0.78 g, 10%), bianthrone A1 [3] (1.1 g, 14%), acetylvismione D, C₂₇H₃₂O₆ (2.5 g, 32%), vismione D [2] (2.5 g, 32%) and

vismione F [3] (0.16 g, 2%). The known compounds were identified by comparison with authentic samples (spectral data, co-TLC and mmp). The new product was purified on a short (to avoid transformations) column of silica gel with CH₂Cl₂.

Acetylvismione D (1), $C_{27}H_{32}O_6$ (M, 452), mp 65-66° (hexane); [α] $_{0}^{22}$ = +6° (0.3, CHCl₃); UV $\lambda_{\text{max}}^{\text{HeCH}}$ nm (log e): 278 (4.68), 320 (3.87), 332 (3.79), 400 (4.09); UV $\lambda_{\text{max}}^{\text{MeOH}}$: 274, 319, 334, 396; $\lambda_{\text{max}}^{\text{MeOH}}$ + AcONa: 274, 319, 334, 396; $\lambda_{\text{max}}^{\text{MeOH}}$ + AlCl₃: 269, 290, 352, 455; MeOH + AlCl₄: 262, 262, 243. $\lambda_{\text{max}}^{\text{MeOH} + \text{AlCl}_3/\text{HCl}}$: 260, 285, 344, 444; IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3380, 1723, 1622; ¹H NMR (CD₃COCD₃): δ 15.98 (1H, s (br), exchg D₂O, 9-OH), 9.56 (1H, s, exchg D₂O, 1-OH), 6.77 (1H, s (br), H-10), 6.55 (1H, d, J = 2 Hz, H-4), 6.30 (1H, d, J = 2 Hz, H-2), 5.45 (1H, t, J)= 7 Hz, = CH, 5.08 (1 H, m (br), = CH), 4.60 (2 H, d, J = 7 Hz, OCH_2), 3.60 + 3.00 (1H + 1H, d + d, J = 16 Hz, 5-CH₂), 3.21 $+2.87 (1H+1H, d+d, J=18 Hz, 7-CH_2), 2.30-1.90 (4H, m, 2)$ \times CH₂), 1.75 (6H, s, COMe + Me), 1.65 + 1.60 (6H + 3H, s + s, 3 \times Me); MS m/z (rel. int.): 452 [M]⁺ (1), 392 [M - AcOH]⁺ (6), 323 $[M - C_5H_9]^+$ (1), 316 $[M - C_{10}H_{16}]^+$ (1), 307 (2), 269 (3), $256 [M - AcOH - C_{10}H_{16}]^+$ (100), 241 (13), 228 (6), 227 (7), 213 (8), 211 (1), 210 (2), 152 (3), 136 (3), 93 (11), 81 (12), 69 (31). (Found: 452.2204. Calc. for C₂₇H₃₂O₆: 452.2199; Found: 392.1992. Calc. for C₂₅H₂₈O₄: 392.1987; Found: 256.0725. Calc. for C₁₅H₁₂O₄: 256.0736.)

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^{*}Part 4 in the series: "Chemistry of *Psorospermum* Genus". For part 3 see Delle Monache, F., Botta, B., Delle Monache, G. and Marini Bettolo, G. B. (1985) *Phytochemistry* 24, 1855.

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